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TWO DIMENSIONAL NMR STUDY OF SOME NATURAL COUMARINS

Key Words: Coumarin, NMR, HETCOR, *Diosmeae*, *Rutinae*

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ABSTRACT

Two dimensional HETCOR and long range HETCOR experiments have been performed on the natural coumarins hopeyhopin, dehydrogeijerin, 6-methoxy-7-(3',3'-dimethylallyloxy)-coumarin, puberulin, oxypeucedanin and isoimperatorin which allowed complete ¹H and ¹³C spectral assignments to be made.

INTRODUCTION

Recently ¹³C data on some natural coumarins and coumarin derivatives¹ have been reported. In the course of our phytochemical studies on the *Diosmeae* and *Rutinae* (sub-family *Rutoideae*) of Southern Africa we

have isolated and identified a number of coumarin structures.^{2,3} We report here ¹H and ¹³C chemical shifts for the compounds hopeyhopin (**1**), dehydrogeijerin (**2**), 6-methoxy-7-(3',3'-dimethylallyloxy)-coumarin (**3**), puberulin (**4**), oxypeucedanin (**5**) and isoimperatorin (**6**). Two dimensional n.m.r. techniques have been used to make unambiguous assignments. The structures of these compounds are given in Fig. 1.

RESULTS AND DISCUSSION

The proton spectrum of hopeyhopin(**1**) is shown in Fig. 2. The assignments of protons 3, 4, 5 and 8 follow from standard chemical shift arguments but are confirmed by ¹H-¹³C correlations. The H-2' resonance is virtually coincident with the OMe resonance as shown by an enlargement of this spectral region, but also by the cross peaks shown in the HETCOR⁴ spectrum.

The ¹³C spectrum of **1** is shown in Fig. 2. The assignment of protonated carbons followed directly from the HETCOR experiment (Fig. 2). Note that in general there is a linear relationship between the proton and carbon chemical shifts. This relationship, however, can only be used for assignment when there is a large difference in chemical shift. For example H-8 is downfield of H-3 while C-8 is upfield of C-3. Similarly the order of H-4, H-5 and C-4, C-5 chemical shifts are not as expected.

In order to assign the quaternary carbons the long range HETCOR⁵ ($J_{CH} = 6$ Hz) pulse sequence was used (Fig. 3). While some of the one bond correlations are still present many three bond C-H correlations are now apparent. H-5 shows cross peaks with C-5, C-4 and C-1', H-4 is correlated

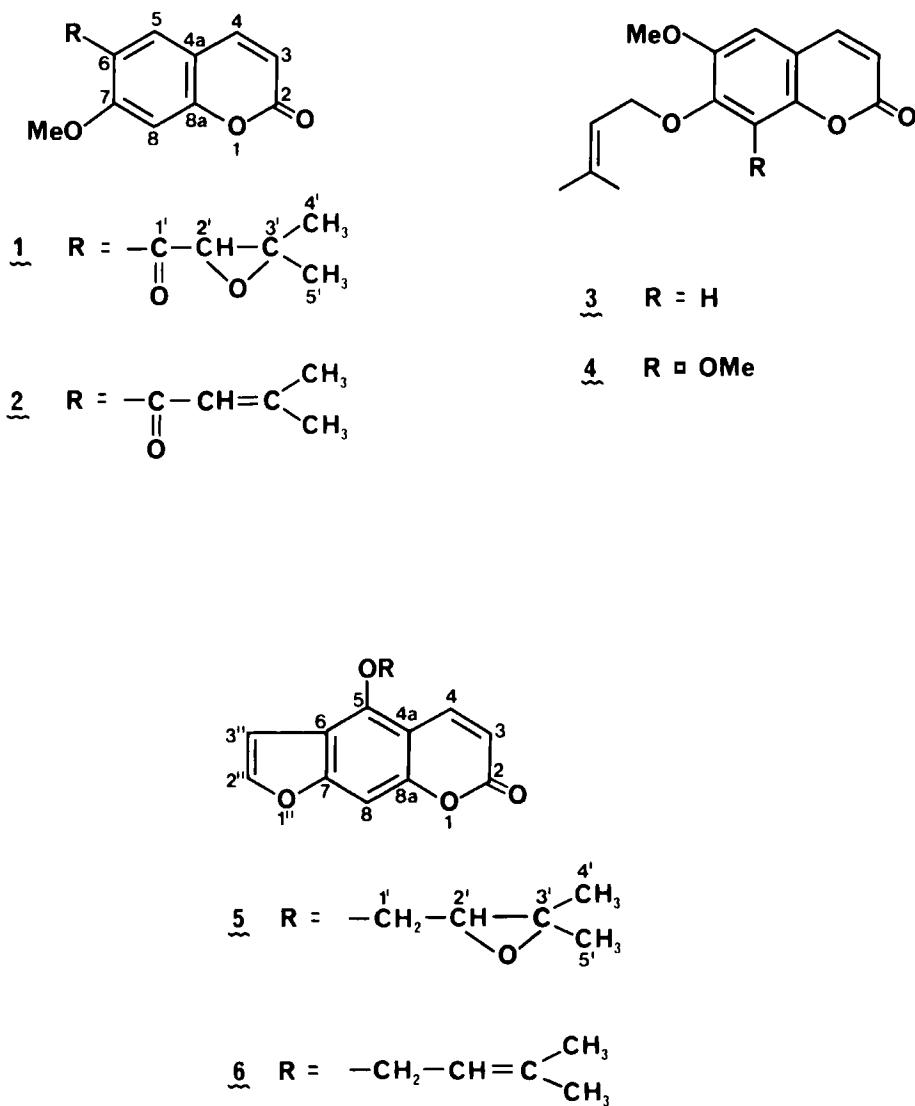


FIG. 1. Structures of compounds 1–6.

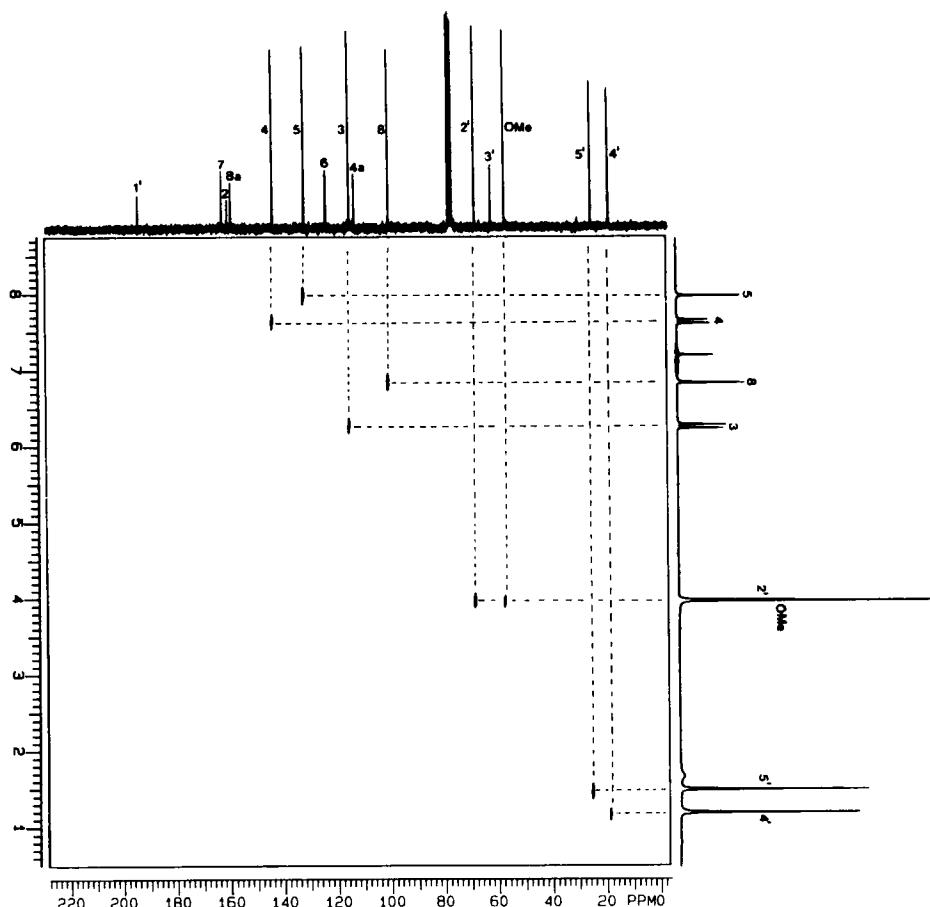


FIG. 2. Two dimensional heteronuclear shift correlated spectrum (HETCOR) of **1** using $J_{C,H} = 140$ Hz.

to C-5, C-4 and C-8a, H-8 is correlated to C-8 and C-6, while H-3 is correlated to C-4a, C-3 and C-2. In addition correlation between OMe and C-7 and H-2' and C-1' confirms the assignment of the other carbon resonances. Reversal of any of the proton assignments would lead to an inconsistent set of ^{13}C assignments.

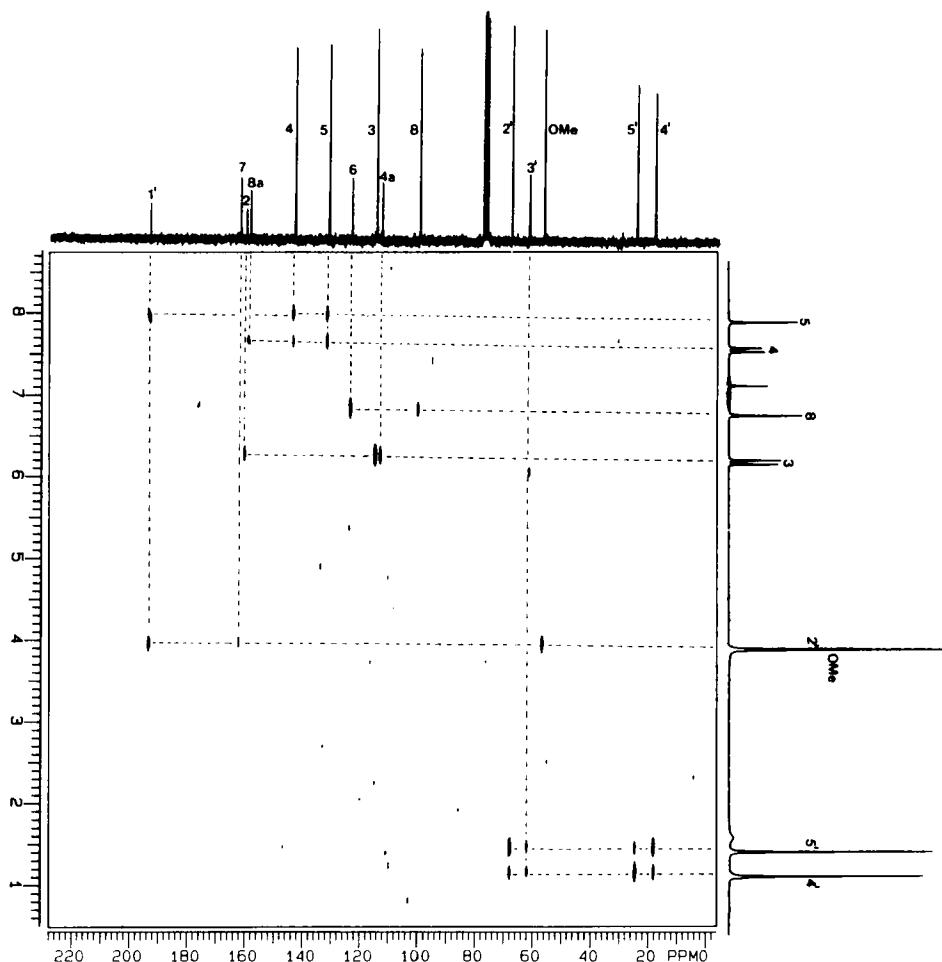


FIG. 3. Two dimensional long range HETCOR spectrum of 1, $J_{C,H} = 6$ Hz.

TABLE 1

¹³C N.M.R. Chemical Shifts for Compounds 1 - 6.

C	Chemical shift (ppm)					
	1	2	3	4	5	6
2	159.9	157.4	161.3	160.6	161.0	161.3
3	114.6	113.9	113.0	115.1	113.0	112.5
4	143.1	143.3	143.3	143.5	138.9	139.6
4a	112.7	112.1	111.1	114.4	107.3	107.4
5	131.3	130.3	107.8	103.6	148.3	148.9
6	123.3	128.2	146.4	144.9	114.0	114.1
7	162.0	160.8	151.9	150.7	157.9	158.1
8	99.8	99.6	100.8	143.0	94.7	94.2
8a	158.7	156.7	149.6	141.8	152.4	152.6
1'	193.4	180.0	66.0	70.3	72.2	69.7
2'	67.9	124.8	118.4	120.0	61.0	119.0
3'	61.8	133.8	138.8	139.3	58.3	139.8
4'	24.6	28.1	25.7	25.8	24.5	25.2
5'	18.2	21.4	18.2	17.9	18.9	18.2
2"					145.2	144.9
3"					104.4	105.0
6-OMe			56.1	56.3		
8-OMe				61.7		
7-OMe		52.2				

TABLE 2
¹H N.m.r. chemical shifts and coupling constants for compounds 1 - 6.

Proton	Chemical Shift (ppm)					
	1	2	3	4	5	6
3	6.29, <i>d</i> (<i>J</i> =9.52)	6.30, <i>d</i> (<i>J</i> =9.53)	6.24, <i>d</i> (<i>J</i> =10.0)	6.31, <i>d</i> (<i>J</i> =9.60)	6.22, <i>d</i> (<i>J</i> =9.8)	6.27, <i>d</i> (<i>J</i> =9.6)
4	7.67, <i>d</i> (<i>J</i> =9.68)	7.67, <i>d</i> (<i>J</i> =9.53)	7.60, <i>d</i> (<i>J</i> =10.0)	7.58, <i>d</i> (<i>J</i> =9.40)	8.13, <i>d</i> (<i>J</i> =9.8)	8.16, <i>d</i> (<i>J</i> =9.6)
5	8.01	7.74	6.81	6.63		
8	7.23	6.85	6.81		7.09	7.15
1'			4.64, <i>d</i> (<i>J</i> =6.1)	4.61, <i>d</i> (<i>J</i> =8.00)	4.61, <i>dd</i> (<i>J</i> =11, 4.4)	4.91, <i>d</i> (<i>J</i> =2.8)
2'	4.01	6.61	5.47, <i>t</i> (<i>J</i> =6.1)	5.55, <i>t</i> (<i>J</i> =8.00)	4.44, <i>dd</i> (<i>J</i> =11, 6.5)	
4'	1.52	2.24*	1.76*	1.74*	3.24, <i>dd</i> (<i>J</i> =6.5, 4.4)	5.53, <i>t</i> (<i>J</i> =8.0)
5'	1.22	2.00*	1.76*	1.68*	1.37*	1.78
2"					1.30*	1.67
3"					7.55, <i>d</i> (<i>J</i> =2.7)	7.59, <i>d</i> (<i>J</i> =2.8)
6-OMe					6.92, <i>d</i> (<i>J</i> =2.7)	6.94, <i>d</i> (<i>J</i> =2.8)
8-OMe					3.88	3.90
7-OMe					4.00	4.00
					3.95	

* Assignments may be reversed.

From the data available it was not possible to assign unambiguously the methyl groups 4' and 5'. To this end difference nOe spectra⁶ were recorded. NOe was detected between H-2' and both H-4' and H-5'. However, based on the size of the nOe, H-5' is *cis* to H-2' and the assignments are as shown. The final set of chemical shifts and assignments are given in Tables 1 and 2.

In a similar manner complete ¹³C and ¹H n.m.r. assignments of compounds 2 to 6 were made. Compound 5 has been studied before and our assignments are in agreement with the literature.¹ Based on a comparison of structurally related coumarins Patra and Mitra¹ propose that the assignment of C-4a and C-8a be interchanged. Using chemical shift correlations we are able to confirm this proposal.

Inspection of Table 1 reveals that, with the exception of substitution at the 5 position, the chemical shift of C-4, C-3 and C-2 are independent of substitution at the phenyl ring. Substitution at the 5 position results in a 4 ppm, highfield shift, of the C-4 resonance. Similarly the C-4a resonance remains invarient while the C-8a resonance shifts 17 ppm.

EXPERIMENTAL

Ground dried leaves and stems were extracted with petrol (60-80 °C). The solvent was removed under reduced pressure and the residue chromatographed on an alumina column using CHCl₃ as solvent. Mixtures of coumarins were separated by preparative TLC (silica gel: solvent C₆H₁₂-EtOAc, 6:4) and recrystallised.

N.m.r spectra were recorded as CDCl_3 solutions in 5 mm sample tubes on a Varian VXR200 spectrometer. Chemical shifts are relative to deuterochloroform (δ_{H} 7.24, δ_{C} 77.0).

Two dimensional HETCOR spectra were recorded using 128 increments, a sweep width of 11668 Hz and an acquisition time of 0.088 s. The final data matrix of dimensions 2048 x 256 was used with apodization functions RE = 0.005, AF = 0.022 and RE2 = 0.005, AF2 = 0.019 respectively. For the standard HETCOR pulse sequence a C-H coupling of 140 Hz was used, while for the long range HETCOR pulse sequence an additional J_{CH} of 6 Hz was used.

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